# A Signal-Amplified Piezoelectric Sensor for the Detection of hs-CRP Using HRP Doped Magnetic Core-Shell Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au Nanostructures as Labels

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A highly sensitive and signal-amplified electrochemical immunoassay for high sensitivity C reaction protein (hs-CRP) has been developed based on an enzyme-catalyzed amplification due to the formation of an insoluble precipitate on the surface of a piezoelectric electrode. A novel horseradish peroxidase (HRP) and CRP antibody co-immobilized magnetic core-shell Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au nanostructures (Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP-HRP) was synthesized and employed as labels for detection of CRP. After sandwich-type immunoreactions between CRP and labels on 96-well micro-titer plate, the immunocomplex on the bottom was subsequently exposed to 3-Amino-9-ethylcarbazole (AEC). The HRP labeled on labels can biocatalyze AEC in the presence of H<sub>2</sub>O<sub>2</sub>, resulting in an insoluble product. When the product solution was dropped on the piezoelectric electrode's surface, it can achieve an obviously decreased of frequency( $\Delta$ f). The amount of precipitation was found proportional to CRP within a certain range of concentration from 0.01 to 200 ng mL<sup>-1</sup> with a low detection limit of 3 pg mL<sup>-1</sup>. Comparing with the conventional ELISA, the immunoassay with the proposed amplified procedure shows improved sensitivity due to the significantly elevated amount of HRP labeled which can significantly amplify the subsequent enzyme catalysis reaction. It can be used to quantitatively determine other biomarker in the sample of human serum if changed the detection antibody.

**Keywords:** Magnetic core-shell Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au nanostructures, hs-CRP, piezoelectric sensor, electrochemical immunoassay

## **1. INTRODUCTION**

The development of new methods for accurate and sensitive detection of disease-related proteins, such as high sensitivity C-reactiveprotein(hs-CRP) in cardiovascular disease, remains a challenge to the scientists in various fields [1,2]. In recent years, some biosensors have been reported as alternatives to the conventional methods, which are reliable but time-consuming or complicated, for rapid detection of biomarkers for cardiovascular disease[3]. Enzyme-linked immunosorbent assay (ELISA) is the most widely used in clinical laboratory However, the concentrations of tumor related proteins are very low at early stage of cancers, which is often beyond the detection limit of ELISA[4]. Thus, new methods that can monitor tumor related proteins of low concetraion level are highly desired. Electrochemical immunoassays, based on specificity of antigen-antibody interactions with electrochemical transducers, have attracted considerable interest due to their intrinsic advantages, such as high sensitivity, simple pretreatment procedure, low cost and fast analysis [5].

The piezoelectric crystal detector can be a very powerful analytical tool because of the relationship shown for the change in frequency to the analyte concentration with high sensitivity. Piezoelectric immunosensors, which are based on a combination of highly specific immuno-recognition and ultra-sensitive QCM mass detection, also offer such opportunities for direct detection of biomarkers[6]. In 1959, Sauerbrey developed an empirical equation for AT-cut quartz crystals vibrating in the thickness shear mode that describes the relationship between the mass of thin metal films deposited on quartz crystals and the corresponding change in resonant frequency of the crystal[7]:

$$\Delta F = -2.3 \times 10^6 F^2 \frac{\Delta M}{A}$$

Where, F = frequency change in oscillating crystal in Hz, F = frequency of piezoelectric quartz crystal in MHz,  $\Box M =$  mass of deposited film in g, and A = area of electrode surface in cm<sup>2</sup>. These relationships not only apply to film deposition but also to particulate deposition. The above explanation shows that the crystal detector indiscriminately changes frequency due to the deposition of mass of any material on its surface. Thus, it is the task of the researcher to choose a coating that will undergo a highly selective immunocomplex binding with the substance to be detected. Thus it can a highly selective sensor be constructed that will be sensitive to the subject to be detected.

Generally, primary or capture antibodies are immobilized onto an AT-cut quartz crystal, adsorption of bacteria onto the immobilized antibodies results in an increase in the surface mass loading of the crystal, and this will decrease the sensor's resonant frequency. According to Sauerbrey equation, the frequency decrease is proportional to the mass change, which in turn is linked to the biomarkers' concentration. A number of piezoelectric immunosensors have been developed for rapid detection of biomarkers including CEA, AFP etc [8,9]. and this has been summarized in several reviews. However, the usage of piezoelectric immunosensors for CRP are rarely reported.

Antibody immobilization is vital in successful development of a piezoelectric immunosensor, and the present immobilization methods are mainly based on silanized layer,SEM layer etc. However,

all the above methods are quite complex for the immobilization of antibody on the electrode which restrict its real application. So if the antibody modification procedure can be excluded, there will be a good prospect of application of piezoelectric method. Moreover, some of the color reaction product in sandwich ELISA was precipitation, so it can be directly measured on QCM electrode[10-13].

Itamar Willner[8have used a series of dye substrates such as o-phenylenediamine (AEC), 3,3'diaminobenzidine(DAB), in immunocytochemistry(IHC) for immunoassay by piezoelectric method. Enzyme labels change their substrates into enzymatic reaction products that can be easily detected either by their electrochemical properties. The precipitation of an insoluble product of AEC or DAB on a tricomponent enzyme-layered electrode has been used in the development of a new electrochemical biosensor to detect acetylcholine [14]. The insoluble enzyme-catalyzed precipitates of immunoreacted complexes which accumulated on the transducers can be used to develop QCM sensors. QCM analyses were then used to probe the precipitation of the insoluble product on the transducer. AEC (3-Amino-9-EthylCarbazole) is one of these reagents, which can produce a positive immunoreaction precipitation product after exposure to HRP enzyme conjugated to an antibody reagent in IHC. Its immunocytochemical visualization of specific antigens of interest can induce a more specific stain by DAB in IHC . So, in the article, ACE was used for a diagnosis.

Nevertheless, this method does not sufficiently meet the sensitivity requirements to determine CRP for clinical analysis, particularly in the early stages of cardiovascular disease. It's because in this kind of electrochemical immunoassays, HRP labeled on a signal tags catalyzed the oxidation of dye substrates by H<sub>2</sub>O<sub>2</sub>, generating the precipitation product. While the labeled amount of HRP enzyme on each secondary antibody of conventional ELISA was limited(only about one HRP in one HRP linked antibody), restricting higher detection sensitivity. Therefore, in order to improve the sensitivity, signal amplification strategy for mass detection techniques by enhancing the HRP label amount in signal tags have been shown to be an effective method.

Nanotechnology-based immunoassays are rapidly emerging as convenient and high sensitive tools for a variety of immune assays [15-22]. Among them, immunomagnetic beads have been the focus of researchers because of their special magnetic properties enable the immunoconjugates to be readily separated from reaction mixtures by means of an external magnetic field [16]. Fe<sub>3</sub>O<sub>4</sub> is one of the most extensively studied magnetic nanomaterials due to its good bio-compatibility, superparamagnetism, low toxicity and easy prepation. Unfortunately, iron nanostructures with a large surface area are easily oxidized and also react between themselves to form aggregates [17]. To avoid such problems, magnetic core-shell nanostructures with  $Fe_3O_4$  as core and silicon oxide(SiO<sub>2</sub>) as shell is proposed. The shell coating can make naked core particles stable, more easily dispersed and tune their properties [18].Gold nanostructures (Au NPs) can provide a natural environment for bimolecular immobilization and facilitate the electron-transfer of biosensor because of their high surface area, excellent electrochemical properties and good biocompatibility [19-22]. Therefore, if Au NPs can be immobilized on the magnetic core-shell nanostructures Fe<sub>3</sub>O<sub>4</sub> @SiO<sub>2</sub> have stimulated a vast amount of research due to its fascinating properties. In the article, a new hybrid spherical nanostructure Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au was employed for immobilization of biomoleculars with a size of about 150~200 nm was designed, with a spherical Fe<sub>3</sub>O<sub>4</sub> nanostuctures core surrounded by a thick silica shell and further decorated with gold nanostructures.

In this work, we report an enzyme-catalyzed signal amplification approach for the electrochemical detection of CRP in human serum using a QCM electrode. core-shell  $Fe_3O_4@SiO_2@Au$  nanocomposites were used as the immobilization substrate for the co-conjugation of HRP and CRP secondary antibody (HRP/anti-CRP). In this way, a nano-bioconjugate named  $Fe_3O_4@SiO_2@Au$ -HRP-anti-CRP containing multiple secondary antibodies and HRP molecules was successfully prepared and acted as a detection antibody. The analytical procedure consists of the immunoreaction of the antigen (CRP) with anti-CRP coated on the 96-well microtiter plates followed by binding  $Fe_3O_4@SiO_2@Au$ -HRP-anti-CRP2. The formed HRP-labeled immunoconjugate showed good enzymatic activity for the oxidation of AEC by  $H_2O_2$ , and CRP could be quantitatively detected by the frequency changing response of the precipitation product of AEC. The usage of  $Fe_3O_4@SiO_2@Au$ -HRP-anti-CRP as a detection antibody with a high HRP label amount can greatly enhance the sensitivity. In addition, this strategy combines the merits of electrochemical method and traditional ELISA method, and smoothes away the cumbersome electrode modification process in electrochemical method and low sensitivity in ELISA.

## 2. EXPERIMENTS

#### 2.1. Chemicals and Materials

HRP-labeled monoclonal mouse anti-human CRP and CRP test kit were obtained from Sigma. Santa Cruze Co.Lte (USA). 3-Mercaptopropyltriethoxysilane (3-MPTES), horseradish peroxidase (HRP EC 1.11.1.7, RZ>3.0, A>250 U/mg) were purchased from Aldrich(USA).Tetraethyl orthosilicate(TEOS) , Glycidoxypropyltrimethoxysilane (GPTES) , 3-Aminopropylt riethoxysilane(APTES), Hydrogen tetrachloroaurate (III) tetrahydrate (HAuCl<sub>4</sub>·4H<sub>2</sub>O), hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) and bovine serum albumin (BSA) were obtained from Sinopharm Group Chem. Re. Co, Ltd. (Shanghai, China). All reagents used were of analytical-reagent grade, and all solutions were prepared with double-deionized water. Phosphate buffer solution (PBS, 0.1M) of various pHs were prepared by mixing the stock solutions of NaH<sub>2</sub>PO<sub>4</sub> and Na<sub>2</sub>HPO<sub>4</sub>, and then the pH was adjusted with 0.1 M NaOH and H<sub>3</sub>PO<sub>4</sub>.

#### 2.2. Apparatus

Electrochemical quartz crystal microbalance (CHI-440A, Shanghai Chen hua instruments Ltd), gold-plated 8 MHz quartz oscillator (Crystal diameter of 13.6mm, gold electrodes 5.3 mm diameter, Shanghai Chen hua instruments Ltd). The morphology of the nanostructures was characterized by an H-7650 transmission electron microscope (Hitachi, Japan). UV-Vis spectra were recorded with a TU-1901 spectrophotometer (Beijing Purkinje General Instrument Co, China). Electrochemical immunoassay measurements were performed on a CHI 660 electrochemical analyzer (CHI, USA) with a conventional three-electrode system comprising a platinum wire as the auxiliary electrode, a

saturated calomel electrode (SCE) as the reference and a glass carbon electrode (GCE) as the working electrode.

#### 2.3. Preparation of core-shell Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au nanostructures

To prepare the composite core-shell nanostructures  $Fe_3O_4@SiO_2@Au$ ,  $Fe_3O_4@SiO_2$ nanostructures were first prepared according to the classical Frens synthesis method [17]. The magnetic nanostructures of  $Fe_3O_4$  ( $Fe_3O_4$  NPs) were synthesized by co-precipitation method. Then, the particles were coated by TEOS (300 µL) and ammonium hydroxide (500 µL) under vigorous stirring in order to the following deposition of the SiO<sub>2</sub> shell. The reaction mixture was heated to 60 °C and kept for 6 h under N<sub>2</sub> atmosphere. The resulting precipitate was separated by magnetic separation and washed for three times with double-deionized water and ethanol respectively to give  $Fe_3O_4@SiO_2$  NPs. The thickness of SiO<sub>2</sub> shell was increased by the hydrolysis and condensation of silica alkoxides. Au nanostructures were adsorbed on  $Fe_3O_4@SiO_2$  NPs surface via the interaction between the negative charge on the Au nanostructures which was reductively synthesized by citrate acid and the positive charge of  $-NH_3^+$  on the SiO<sub>2</sub> shell modified by 3-aminopropyltriethoxysilane (APTES,300 µL)[21]. After magnetic separation, the obtained  $Fe_3O_4@SiO_2@Au$  nanostructures were obtained and rinsed with double-deionized water. The procedures are shown in Figure.1

#### 2.4. Preparation of Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP-HRP labels

The procedure to prepare  $Fe_3O_4@SiO_2@Au$ -HRP-anti CRP signal tags was shown in Figure.1. About 1.0 mL of  $Fe_3O_4@SiO_2@Au$  nanostructures suspension was initially adjusted to pH 8.2 using Na<sub>2</sub>CO<sub>3</sub>, and then 1.0 mL of the original HRP-anti CRP was added into the mixture and incubated for 12 h at 4 °C with slightly stirring. After magnetic separation, the obtained  $Fe_3O_4@SiO_2@Au$ -HRP-anti CRP labels were incubated with 1 mL of 1.0 wt% HRP for 1 h to block the nonspecific sites on the uncovered surface of  $Fe_3O_4@SiO_2@Au$ .



Figure 1. Schematic of the preparation of Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP conjugation.

The synthesized Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au- HRP-anti CRP-HRP signal tag was stored in pH 7.4 PBS at 4  $^{\circ}$ C when not in use.

## 2.5. Immunoassay procedures

A schematic representation of the steps used to perform the electrochemical ELISA was shown in Figure. 2 up. An anti CRP pre-coated polystyrene 96-well microtiter plate was incubated with 100  $\mu$ L of different concentrations CRP for 30 min at 37 °C. After incubation, the wells were washed six times with pH 7.4 PBS containing 0.05% (W/V) Tween-20. Then 100  $\mu$ L of Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRPanti CRP conjugations (1:10 dilution with PBS containing 1 wt.% BSA) were pipetted into each well and incubated at 37 °C for 30 min. After the wells were rinsed, 100  $\mu$ L of 1mmol/L AEC and H<sub>2</sub>O<sub>2</sub> mixture solution were added to each well, and the enzymatic reaction was allowed to proceed for 10 min at 37 °C. 20 $\mu$ L of the resulting reaction solution was transferred and dropped onto QCM electrode with 5  $\mu$ L PBS (pH 7.0,Figure.2 down).



Figure 2. Schematic of the preparation of  $Fe_3O_4@SiO_2@Au$ -HRP-anti CRP-HRP conjugation and the electrochemical enzyme-linked immunoassay (ELISA) procedure with the application of  $Fe_3O_4@SiO_2@Au$ -HRP-anti CRP-HRP nano conjunctures as the signal enhancer(up) and the piezoelectric electrode(a) and detection system(b).

## **3. RESULTS AND DISCUSSION**

## 3.1. Characterization of Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP nano-bioconjugation

The core-shell  $Fe_3O_4@SiO_2@Au$  nanostructures was used to label HRP-anti CRP because its high specific surface area may enhance the immobilization amount of HRP-anti CRP and HRP, and thus a nano-bioconjugation with mutiple anti CRP sites and HRP molecules can be obtained, which favors the HRP enzymatic oxidation of AEC by  $H_2O_2$  and hence provides amplified signal in the electrochemical ELISA.

Transmission electron microscopy(TEM) images showed that both  $Fe_3O_4@SiO_2$  and  $Fe_3O_4@SiO_2@Au$  nanostructures were of well spherical structure and preferable mono disperisity in size. The average diameter of  $Fe_3O_4$  nanostructures and core-shell  $Fe_3O_4@SiO_2@Au$  nanostructures were about 100 nm (Figure. 3a) and 120 nm (Figure. 3b), respectively. Upon deposition of gold shell to  $Fe_3O_4@SiO_2$  nanostructures, the diameter of the  $Fe_3O_4@SiO_2$  particle increased by about 20 nm, demonstrating that the Au shell was about 10 nm thick. As shown in Figure. 3c,  $Fe_3O_4@SiO_2@Au$ -HRP-anti CRP-HRP suspension was homogeneously dispersed in the solution. Once an external magnetic field was applied, the  $Fe_3O_4@SiO_2@Au$ -HRP-anti CRP-HRP was attracted quickly toward the magnet, leaving the bulk solution clear and transparent, indicating the magnetic activity of the  $Fe_3O_4@SiO_2@Au$ -HRP-anti CRP-HRP (Figure. 3d).



**Figure 3.** TEM images of (a) Fe<sub>3</sub>O<sub>4</sub>@SiO, (b) Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au (c) Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au -HRP-anti CRP (d) c in absence (right) and presence (left) of external magnetic field.

The XRD patterns were applied on characterization of the Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub> nanostructures and Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub> @Au nanostructures. The diffraction peak positions at  $2\theta = 30.0^{\circ}$ ,  $35.3^{\circ}$ ,  $42.6^{\circ}$ ,  $53.4^{\circ}$ , and  $62.5^{\circ}$  can be attributed to the (220), (311), (400), (422), (511), and (440) planes of Fe<sub>3</sub>O<sub>4</sub> in a cubic phase , respectively. While in Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub> , @Au, the diffraction peak positions at  $2\theta = 38.1^{\circ}$ ,  $44.2^{\circ}$ ,  $64.5^{\circ}$ ,  $77.5^{\circ}$ , and  $81.6^{\circ}$  can be attributed to the (111), (200), (220), (311), and (222) planes of Au, respectively. Other peaks can be attributed to Fe<sub>3</sub>O<sub>4</sub>. There is no crystal peak of SiO<sub>2</sub>, it's because SiO<sub>2</sub> existed in the presence of amorphous. The simultaneous observation of both Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub> and gold peaks verified successful assembly of Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub> @Au nanostructures.

The magnetic properties of the nanostructures are illustrated in Figure 4. It can be seen that a small coercivity or remanence existed at room temperature, indicating the  $Fe_3O_4@SiO_2$  and  $Fe_3O_4@SiO_2$  @Au have super paramagnetic properties. As a result, the  $Fe_3O_4@SiO_2$  @Au could be isolated from solution by an adscititious magnet. It can also be seen that the saturation magnetization for  $Fe_3O_4@SiO_2$  @Au is smaller than that for  $Fe_3O_4@SiO_2$ .The saturation magnetization values were 2.68 emu/g and 1.79 emu/g for  $Fe_3O_4@SiO_2$  and  $Fe_3O_4@SiO_2$  @Au nanostructures, respectively.



**Figure 4.** Hysteresis loops of magnetic silica nanoshperes with (a)Fe<sub>3</sub>O4@SiO<sub>2</sub> and (b) Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub> @Au at 300K,respectively

UV-Vis absorption spectrometry was also employed to characterize the nano-bioconjugation of  $Fe_3O_4@SiO_2@Au$ -HRP-anti CRP. As could be seen, the  $Fe_3O_4$  nanostructures showed no characteristic absorption in the examined range (Figure. 5a). When gold nanostructures were deposited on the surface of nano  $Fe_3O_4$ , a new absorption band centered at 520 nm resulted from gold nanostructures was observed (Figure. 5b), indicating the  $Fe_3O_4@SiO_2@Au$  nanostructures were successfully prepared, which was in good agreement with the results of Qiu et al[17]. After the HRP-anti CRP(Figure.5c) molecules being labeled onto the surface of the  $Fe_3O_4@SiO_2@Au$  nanostructures, the obtained  $Fe_3O_4@SiO_2@Au$ -HRP-anti CRP nano-bioconjugate exhabited two absorption peaks at 280 and 410 nm(Figure.5d). The peak at 520 nm originated in Au peaks on the  $Fe_3O_4@SiO_2@Au$ 

nanostructures, and the peak at 280 nm and 410nm was attributed to the absorption of HRP-anti CRP proteins as judged from Figure. 4c. On the basis of the above results, it can be concluded that  $Fe_3O_4@SiO_2@Au$ -HRP-anti CRP nano-bioconjugation was successfully prepared and could be used in the electrochemical ELISA of CRP.



**Figure 5.** UV-Vis absorption spectra of (a) Fe<sub>3</sub>O<sub>4</sub> nanostructures, (b) Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au nanostructures, (c) HRP-anti CRP, (d) Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP conjugation.

#### 3.2. The characterization of signal amplification by different labels

To clarify the advantages of the QCM methods using Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP-HRP bioconjugation as the detection antibodies, we constructed two types of detection antibodies as labels including HRP-anti CRP, and Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP. 5 ng/mL of CRP was used as an example for the evaluation of the electrochemical signal response. The judgment was based on the current change slope ( $\Delta f$ ) between QCM data of the oxidation product of AEC generated in the electrochemical immunosensor and the background current of QCM(Figure.5, dotted line). As shown in Figure. 5, the use of Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP as the detection antibody ( $\Delta f$ =22.1 Hz/logC A, Figure. 6a) exhibited much higher current response than that of applying HRP- anti CRP ( $\Delta f$  =6.4 Hz/ logc, Figure. 6b). The former is about 3.5 folds of the latter. This showed that the immunosensor by Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP-HRP can generate significant signal amplification for the development of ultrasensitive amperometic immunosesay method than HRP-anti CRP. The reason might be the fact that binding amount of HRP is much more in Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP-HRP than HRP- anti CRP. When one Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP-HRP detection antibody reacted with the antigen. Other HRPs can also participate the enzymatic reaction, which facilitated the

oxidation of AEC by  $H_2O_2$  to produce more precipitation product for signal amplification and improved sensitivity.



**Figure 6.** The calibration curve of the immunosensor modified with Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP (a),HRP-anti CRP (b) immunocomplexes.

#### 3.3. Optimization of experimental conditions

When using the Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP-HRP nano-bioconjugate as a detection antibody, the ratio of HRP-anti CRP and HRP on the surface of Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au would have important effect on the amount of HRP, which determined the interaction efficiency. To determine optimal ratio of HRP and anti CRP, various volume ratios of 1 mg·mL<sup>-1</sup> HRP and original anti CRP including 20:1~300:1, were used for the preparation of Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP-HRP. After incubated with 10 ng·mL<sup>-1</sup> CRP, the immunoconjugates were allowed to react with HRP substrate AEC. The reaction product was measured by QCM value increases with increased to the plateau at the ratio of 1:100(Figure.7a). Thus, HRP to anti CRP 1:100 was chosen as the optimal volume ratio to prepare Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP-HRP.

The concentrations of  $H_2O_2$  and AEC are important parameters that influence the activity of HRP. Using 10.0 ng·mL<sup>-1</sup> CRP as a model, we monitored the effect of concentrations of  $H_2O_2$  and AEC on the current response of the electrochemical ELISA. The reduction peak current of the enzymatic oxidation product increased with the increasing the concentrations of  $H_2O_2$  (Figure. 7b) and AEC(Figure.7c) maintained the maximum value at higher concentrations. Afterward, the enzymatic reaction rate depended on the amount of the labeled HRP. Therefore, the optimal concentrations of 9.8 mmol/L for AEC and 2.33 mmol/L for  $H_2O_2$  were used for the immunosensor.



**Figure 7.** (a) Optimization of volume ratio of HRP to HRP-anti CRP for preparation of Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>@Au-HRP-anti CRP. Effects of concentration of AEC(b), H<sub>2</sub>O<sub>2</sub>(c), and pH of PBS on cyclic voltammetric response at the immunosensor(d).

Incubation time and incubation temperature also affected the performance of the activity of enzyme(Fig 7d). The optimum incubation temperature of antigen-antibody reaction should be close to the normal temperature of human body (37 °C), so all the experiments were carried out at 37 °C. At this temperature, the enzymatic reaction equilibrium can be achieved in 30 min. So the optimal reaction time was selected as 30 min.

## 3.4. Analytical performance

Under the optimal conditions, the electrochemical ELISA was carried out to analysis various concentrations of CRP standard solution and the CV response was recorded. The  $\Delta f$  of QCM response increased with the increasing of CRP concentration. The linear range was from 0.01~200 ng·mL<sup>-1</sup>. The linear regression equation was  $\Delta f(\text{Hz}) = 22.1 \log C_{\text{CRP}}(\text{ng·mL}^{-1}) + 0.12$  with a coefficient of 0.9923, it's detection range was between 0.01 to 200 ng mL<sup>-1</sup> with a limit of detection(LOD) of 5 pg·mL<sup>-1</sup>.

To highlight the advantages of the  $Fe_3O_4@SiO_2@Au$  nanoparticles label based electrochemical ELISA for CRP detection, the analytical properties of the developed immunoassay were compared with those of other CRP immunoassay reported previously. The results are listed in Table 2. It is

obvious that the linear range was broader and the LOD was comparably low. This result indicated that use of  $Fe_3O_4@SiO_2@Au$  nanoparticles as labels could obviously improve the sensitivity and dynamic range of the immunoassay.

Immunoassay	Linear range (ng•mL <sup>-1</sup> )	LOD (ng·mL <sup>-1</sup> )	Detection antibody	Ref.
ELISA	300-10000	100	HRP -labeled	[23]
			antibody	
Electrochemistry	6.6-66000	2.1	silicon nanowire	[24]
			array-antibody	
SPR-based	2000-5000	650	CRP-antibody	[25]
immunosensor				
Electrochemiluni	100 -10000	100	CRP Ab2 labeled on	[26]
nescence			liposomes	
			containing	
			$Ru(bpy)_3^{2+}$	
	10-10000	3.2	$[\operatorname{Ru}(\operatorname{bpy})_3]^{2+}$	[27]
			containing	
			polystyrene	
			microspheres	
This method	0.01-200	0.003	Fe <sub>3</sub> O <sub>4</sub> @SiO <sub>2</sub> @Au-	This work
			HRP-antibody	

Table 1	1.0	Comparison	of anal	ytical	propert	ies of v	various	CRP	immunosensors	and	immunoassay	ys
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# 3.5. Specificity, reproducibility and stability

To evaluate the specificity of the proposed electrochemical ELISA for CRP detection, various biomarkers including carcinoma antigen 125 (CA 125), carcinoembryonic antigen (CEA), human IgG (HIgG), and prostate-specific antigen (PSA) were tested. The electrochemical signals were recorded in  $10 \text{ ng} \cdot \text{mL}^{-1}$  CRP with and without the interfering agents. No significant difference of currents was observed in comparison with the result obtained in the presence of only CRP, indicating the specificity of the proposed electrochemical ELISA was acceptable.

The reproducibility of the immunoassay was evaluated by using the coefficients of variation of intra- and inter-assays. Taking 10  $\text{ng} \cdot \text{mL}^{-1}$  CRP for examples, the intra-assays and inter-assays of the electrochemical was evaluated using one electrochemical ELISA for five repeat assays and five copies of electrochemical ELISA for one time assay. The coefficients of variation of the intra-assay and inter-assay were 3.4% and 2.4%, respectively. Thus, the reproducibility of the electrochemical ELISA is satisfactory.

The stability of the synthesized nano-bioconjugate of  $Fe_3O_4@SiO_2@Au$ - HRP-anti CRP was also examined. When not in use, it could be stored in pH 7.0 PBS for at least 2 weeks without obvious signal change. Moreover, it retained 92.4% of its initial response after a storage period of 3 weeks. The slow decrease in the current response may be attributed to the gradual deactivation of the immobilized biomolecules on the surface of the nanostructures.

## 3.6. Determination of CRP in human serum samples

In order to investigate the possibility of the developed method for clinical analysis, some of patient human serum samples were examined by the developed method and the results were compared with the classical ELISA method. The concentration of hs-CRP in three samples were determined to be consistent with the results obtained by the classical ELISA method(Table 2), indicating that the proposed immunoassay could be satisfactorily applied to the clinical determination of hs-CRP.

**Table 2.** Comparison of hs-CRP levels in patient human serum samples (n=3) determined using two methods

Serum samples	The developed method concentration( $ng \cdot mL^{-1}$ )	The classical ELIAS concentration( $ng \cdot mL^{-1}$ )
Sample 1	10.2±0.13	10
Sample 2	5.27±0.12	5.5
Sample 3	21.1±0.22	22

## 4. CONCLUSIONS

In this study, an ultrasensitive electrochemical ELISA method was developed for the detection of CRP by using a newly designed detection antibody. The detection antibody was prepared by loading high-content HRP and HRP-anti CRP antibody on core-shell  $Fe_3O_4@SiO_2@Au$  nanostructures. The signal was greatly amplified to detect hs-CRP comparing with other ELISA method. The results described above demonstrate that amplification proceeds via  $Fe_3O_4@SiO_2@Au$ -HRP-anti CRP binding to a anti CRP/CRP complex, and that the biocatalyzed AEC to produce the precipitation of an insoluble product on the surface of the QCM electrode can significantly extend the sensitivity, leading to an unprecedented detection limit for CRP of 5pg/mL. With a sandwich-type immunoassay format, ultrahigh sensitivity was achieved by the enzymatic signal amplification using AEC as substrate. The use of core-shell  $Fe_3O_4@SiO_2@Au$  nanostructures as signal enhancers not only resulted in better performance, but also favored the rapid separation and purification of detection antibody on magnetic field. The detection limit of the proposed electrochemical ELISA was 100 times lower than that of the traditional ELISA method. The proposed method has a broader linear range, high sensitivity, convenient preparation and low detection limit, which provided a promising potential in clinical diagnosis.

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# References

- 1. R.P. Smith, B.J. Lipworth, A. I. Cree, E.M Spiers, J.H.Winter. Chest. 108 (1995)1288-1291
- 2. S. Black, I. Kushner, D. Samols. J of Biological Chem, 279(2004)48487-48490
- 3. V-L.Inger, M. A. Willem. Biosensors and Bioelectronics. 21 (2006)1141–1148
- 4. G.J. Wolbink, M.C. Brouwer, S. Buysmann, I.J.T. Berge, C.E.Hack. *J of Immunology*. 157(1996)1473-479
- 5. J.G Hou, N Gan, F.T Hu1, L.Zheng, Y.T Cao, T.H Li. Int. J. Electrochem. Sci., 6 (2011) 2845 2858.
- 6. B. Attili, A. Suleiman. Microchem. J. 54(1996)174–179
- 7. G.G. Sauerbrey. Z Phys, 155(1959)206-212
- 8. A. Bardea, E. Katz, I. Willner. Electroanalysis. 12(2000)1097-1106
- 9. T.N. Ermolaeva, E.N. Kalmykova, Russ. Chem. Rev. 75(2006)397-412
- 10. M.Pohanka, O. Pavlis, P. Skladal. Sensors. 7(2007)341-353
- 11. Z.Y. Wu, C.P He, S.P. Wang, G.L. Shen, R.Q. Yu. Chem J of Chin Univs. 22 (2001)542-546
- 12. Z.Z. Chen, K.M. Wang, Y.K. Chen. Chem J of Chin Univ. 23(2002) 1044–1046
- 13. E. Briand, M.Salmain, C. Compere . Biosensors and Bioelectronics, 22(2007) 2884–2890.
- 14. Z.Y. Wu, J. Wu, S.P. Wang, G.L. Shen, R.Q. Yu. *Biosensors and Bioelectronics*, 22(2006) 207–212.
- 15. N.L. Rosi, C.A. Mirkin. Chem. Rev, 105(2004)1547-1562
- 16. F.Lacharme, C.Vandevyver, M.A.M. Gijs. Anal. Chem, 80(2008) 2905–2910
- 17. D. Tang, R.Yuan, Y. Chai. Clin. Chem, 53(2007)1323-1329.
- 18. N.Gan, H.J. Jin, T.H. Li, L. Zheng. Int J of Nanomedicine, 6(2011) 3259-3269
- 19. L.P. Bicelli, B. Bozzini, C. Mele, L. D'Urzo. Int. J. Electrochem. Sci., 3 (2008) 356 408
- 20. X.H. Huang, I.H. El-Sayed, W. Qian, M.A. El-Sayed. J. Am. Chem. Soc, 128(2006) 2115-2120
- 21. M.E. Khosroshahi, L. Ghazanfari. Materials Chemistry and Physics. 133(2012)55-62.
- 22. H.F. Chen, D.P. Tang, B. Zhang, B.Q. Liu, Y.L. Cui . G.N. Chen. Talanta, 91(2012)95-102.
- 23. W.L. Roberts, L. Moulton, T. C. Law, G. Farrow, M. C. Anderson, J. Savory, N. Rifai. *Clinical Chemistry*. 47(2001)418-425
- 24. M.H. Lee, K.N. Lee, S.W. Jung, W.H. Kim, K.S. Shin, W.K. Seong.*Int J Nanomedicine*. 3(2008)117–124.
- 25. H.F.M. Martin, H. Markus, K. Michael. Biosensors and Bioelectronics, 21(2006)1987–1990.
- 26. W Zhan, A. J. Bard. Anal. Chem., 79 (2007) 459-463
- 27. W.J Miao, A.J. Bard. Anal. Chem., 76 (2004)7109-7113

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